

WEST BENGAL STATE UNIVERSITY

B.Sc. Honours 5th Semester supplementary Examination, 2021

CEMACOR12T-CHEMISTRY (CC12)

ORGANIC CHEMISTRY-V

Time Allotted: 2 Hours

Full Marks: 40

The figures in the margin indicate full marks. Candidates should answer in their own words and adhere to the word limit as practicable. All symbols are of usual significance.

Answer any four questions taking one from each unit

UNIT-I

1.	(a)	Describe Fischer indole synthesis of 2-methylindole. Write plausible mechanism. How would you demonstrate which nitrogen is lost during cyclisation?	2+1
	(b)	How could you synthesize anthracene starting from naphthalene using Diels Alder reaction as one of the key steps?	2
	(c)	Arrange furan, pyrrole and thiophene in order of increasing aromaticity.	2
	(d)	Predict the products in the following reaction with plausible mechanism.	3



2. (a) Identify A and B and explain all the steps.



(b) Unlike pyrrole indole undergoes electrophilic substitution at C-3. Explain.
(c) How can you convert toluene into 1,4,6-trimethyl naphthalene?
(d) Compare the basicity of pyrrole and pyridine.

3

UNIT-II

3. (a) Acetolysis of both cis- and trans-tosylate shown below give the same diacetate. Explain.





3

3

2

2

2

2

- (b) The cis-1,2-dimethylcyclohexane is less stable than its trans isomer, but cis-1,3dimethylcyclohexane is more stable than its trans isomer. Draw the most stable conformations of both and explain.
- 4. (a) Compare the rate of hydrolysis of cis and trans isomer of ethyl 4-tbutylcyclohexane carboxylate.
 - (b) Write down the preferred conformation of cis-4-hydroxyl cyclohexane 2+1 carboxylic acid. What happens when it is heated?

UNIT-III

5. (a)) Write down the characteristics of pericyclic reaction.	2
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(b) Write down the product of the following reaction.



- (c) Dimerisation of cyclopentadiene in thermal condition gives preponderantly the *endo* cycloadduct under kinetically controlled conditions. Explain.
- 6. (a) Define stereoselectivity of 4 pi system under photochemical and thermal 2 condition.
 - (b) Write down the product of the following reaction.



(c) What happens when compound A is subjected to heating?



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UNIT-IV

7.	(a)	Write down pyranose structure of D – Glucose.	1
	(b)	How can you selectively methylate C ₃ -OH of D-glucose?	2
	(c)	E, F and G are the three aldohexoses. E and F yield D-sorbitol when they are catalytically hydrogenated. E and F yield different osazones when treated with excess phenyl hydrazine. F and G give the same osazone but different alditols. Give structures of E, F and G assuming that F and G are D-aldohexoses.	3
	(d)	What happens when D- glucose is subjected to HNO ₃ oxidation?	2
8.	(a)	What are epimers?	1
	(b)	Describe mutarotation of glucose.	2
	(c)	An aldpentose [P] can be oxidized with dil HNO ₃ to an optically active aldaric acid. Kiliani- Fisher synthesis starting with [P] gives two new aldoses [Q] and [R]. Aldose [Q] can be oxidized to an optically inactive aldaric acid, but aldose [R] is oxidized to an optically active aldaric acid. Assuming the D-configuration, give the structures of [P], [Q] and [R] and also justify the assignments.	3
	(d)	State with mechanism what happens when D-fructose is heated with Tollen's reagent.	2
		UNIT-V	
9.	(a)	Write down the steps for the synthesis of a tripeptide Gly-Ala-Phe in the solid phase with the help of Merrifield resin.	3
	(b)	Describe Sanger degradation method for N-terminal amino acid determination of peptide.	3
	(c)	Between A-T and G-C, hydrogen bonding in which pair is stronger? Why?	2
	(d)	RNA undergoes alkaline hydrolysis at a faster rate than DNA. Explain.	2
10	.(a)	Describe synthesis of tripeptide Leu-Val -Pro using chemical method.	3
	(b)	Outline the Gabriel synthesis of glycine.	3
	(c)	Briefly explain the factors responsible for the stabilisation of a DNA duplex.	2
	(d)	Differentiate between nucleosides and nucleotides.	2

N.B.: Students have to complete submission of their Answer Scripts through E-mail / Whatsapp to their own respective colleges on the same day / date of examination within 1 hour after end of exam. University / College authorities will not be held responsible for wrong submission (at in proper address). Students are strongly advised not to submit multiple copies of the same answer script.

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